

Enicostemins A and B, New Secoiridoids from *Enicostemma verticillatum*

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Z. Naturforsch. **2011**, *66b*, 749–751; received March 12, 2011

The new secoiridoids Enicostemins A (**1**) and B (**2**) were isolated from the *n*-butanol-soluble fraction of *Enicostemma verticillatum* along with gentiocrucine (**3**) and rutin (**4**), which were isolated for the first time from the genus *Verticillatum*. Their structures were assigned based on spectroscopic studies.

Key words: *Enicostemma verticillatum*, Secoiridoids, Enicostemins A and B

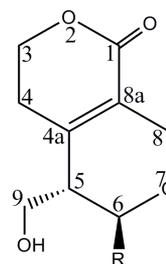
Introduction

The genus *Enicostemma* belonging to the family Gentianaceae comprises four species [1, 2]. One of these is *Enicostemma verticillatum*, which is widely distributed in South America, Africa, and Asia [1]. In Pakistan, it is mainly found in Thatta, Badin, Hyderabad, Mirpur, Gharo, and Manghopir [1]. *E. verticillatum* is a bitter tonic and is used as a substitute for chirayita as a blood purifier. The literature survey revealed that only a flavone *C*-glucoside has so far been reported from this species [3]. Herein, we report the isolation and structure elucidation of two new secoiridoids named as enicostemins A (**1**) and B (**2**) (Fig. 1) along with gentiocrucine (**3**) [4] and rutin (**4**) [5], which have been isolated for the first time from this species.

Results and Discussion

Enicostemin A (**1**) was obtained as a colorless gummy solid. The high-resolution EI-MS of **1** exhibited an $[M]^+$ peak at $m/z = 214.0841$ corresponding to the molecular formula $C_{10}H_{14}O_5$ (calcd. 214.0838). The IR spectrum indicated the presence of hydroxyl groups (3400 cm^{-1}), a conjugated carbonyl function (1701 cm^{-1}) and conjugated double bonds (1635 cm^{-1}). The UV spectrum showed a strong absorption at 240 nm which is characteristic of secoiridoids [6].

The ^1H NMR spectrum showed the multiplet of an oxymethine proton at $\delta_{\text{H}} = 3.69$, and oxymethylene protons were observed at $\delta_{\text{H}} = 4.41$ (m, 1H, H-3), 4.38



1. R = CH_2OH
2. R = *O*- β -D-glucosyl

Fig. 1. Structures of enicostemins A (**1**) and B (**2**).

(m, 1H, H-8), 4.25 (d, $J = 16.0$ Hz, 1H, H-8), 4.02 (m, 1H, H-3), 3.78 (dd, $J = 11.0, 7.0$ Hz, 1H, H-10), 3.71 (m, 2H, H-9), and 3.64 (dd, $J = 11.0, 4.5$ Hz, 1H, H-10).

The broad band (BB) and distortionless enhancement by polarization transfer (DEPT) ^{13}C NMR spectra showed 10 signals comprising 5 methylene, 2 methine and 3 quaternary carbons. The most downfield signal at $\delta_{\text{C}} = 165.4$ (C-1) was attributed to an α,β -unsaturated ester while signals of conjugated olefinic carbons were observed at $\delta_{\text{C}} = 153.8$ (C-4a) and 125.0 (C-8a). The oxymethine carbon gave a signal at $\delta_{\text{C}} = 77.8$ (C-6), and four oxymethylene carbons resonated at $\delta_{\text{C}} = 67.6$ (C-8), 65.9 (C-3), 63.3 (C-10), and 60.5 (C-9). Both the ^1H and ^{13}C NMR data showed close resemblance to those of 5,6-dihydro-5-hydroxymethyl-6-methyl-1*H*,3*H*-pyrano[3,4-*c*]-pyran-1-one [7, 8]. However, the absence of signals due to an olefinic proton indicated the presence

of a tetra-substituted double bond which could be located between C-4a and C-8a. Compound **1** also differs from 5,6-dihydro-5-hydroxymethyl-6-methyl-1*H*,3*H*-pyrano[3,4-*c*]-pyran-1-one in having a hydroxymethyl moiety at C-6 instead of a methyl group. The presence of hydroxymethyl groups at both C-5 and C-6 could also be confirmed through ^1H - ^1H correlated spectroscopy (COSY) as well as heteronuclear multiple-bond correlation spectroscopy (HMBC), as illustrated in Fig. 2. Upon irradiation of the oxymethylene protons of C-9 at $\delta_{\text{H}} = 3.71$ the multiplet of H-5 collapsed into a doublet ($J = 2.7$ Hz). On the other hand, irradiation of H-10_a at $\delta_{\text{H}} = 3.78$ changed the multiplet of H-6 into a double doublet ($J = 2.7$ and 7.0 Hz), and irradiation of H-10_b at $\delta_{\text{H}} = 3.64$ converted the multiplet of H-6 into a double doublet ($J = 2.7$ and 4.5 Hz). The smaller coupling constant between H-5 and H-6 was quite similar to the ones of gentiopicroside [9] and 5,6-dihydro-5-hydroxymethyl-6-methyl-1*H*,3*H*-pyrano[3,4-*c*]-pyran-1-one [7], and therefore both the protons at C-5 and C-6 are equatorial, and the hydroxymethyl substituents at C-5 and C-6 are *trans*-oriented. Conclusive evidence was provided by nuclear Overhauser enhancement spectroscopy (NOESY) which showed a correlation between H-6 and the oxymethylene protons at H-9. Thus enicostemin A (**1**) could be assigned the structure 5,6-bis(hydroxymethyl)-4,5,6,8-tetrahydro-1*H*,3*H*-pyrano [3,4-*c*]pyran-1-one (Fig. 1).

Enicostemin B (**2**) was obtained as a colorless gummy solid. The molecular formula $\text{C}_{15}\text{H}_{22}\text{O}_{10}$ was established by HR-FAB-MS showing an $[\text{M}-\text{H}]^-$ peak at $m/z = 361.1134$ (calcd. for $\text{C}_{15}\text{H}_{21}\text{O}_{10}$, 361.1129). The IR and UV spectra were very similar to those of **1**.

The ^1H NMR spectrum was also similar to that of **1** except for the downfield shift of H-6 to $\delta_{\text{H}} = 5.01$. The anomeric proton was observed at $\delta_{\text{H}} = 5.90$ as a doublet ($J = 7.4$ Hz, H-1'). Further oxymethine protons of the hexose moiety were observed in the range $\delta_{\text{H}} = 3.91 - 3.53$ while the oxymethylene protons were observed at $\delta_{\text{H}} = 3.72$ (dd, $J = 11.0, 5.5$ Hz, 1H, H-6') and 3.62 (dd, $J = 11.0, 4.1$ Hz, 1H, H-6'). The larger coupling constant of the anomeric proton indicated a β -glycosidic linkage. Enzymatic hydrolysis provided an aglycone which could not be worked up due to paucity of material. The glycone could be identified as D-glucose through co-TLC with an authentic sample and the sign of its optical rotation. The downfield shift of C-6 indicated the presence of a β -D-glucopyranosyloxy moiety at this position.

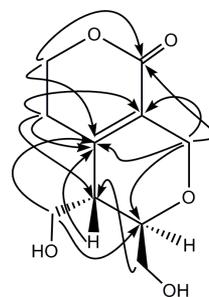


Fig. 2. Important HMBC correlations of enicostemin A (**1**).

The ^{13}C NMR spectrum showed 15 signals comprising 5 methylene, 7 methine and 3 quaternary carbons. It showed common features to those of **1** except for the signal of C-6 being shifted downfield to $\delta_{\text{C}} = 98.7$ (C-6) due to the presence of two vicinal oxygen atoms. In addition, the signals of a hexose moiety were observed at $\delta_{\text{C}} = 103.9$ (C-1'), 78.1 (C-5'), 77.6 (C-3'), 73.0 (C-2'), 71.6 (C-4'), and 62.8 (C-6'). Conclusive evidence was provided by the HMBC spectrum showing a 3J correlation of the anomeric proton at $\delta_{\text{H}} = 5.90$ with C-6 ($\delta_{\text{C}} = 98.7$). Similarly H-6 at $\delta_{\text{H}} = 5.01$ showed a 3J correlation with the anomeric carbon ($\delta_{\text{C}} = 103.9$). Further HMBC and NOESY correlations were similar to those of **1** allowing us to assign the structure of **2** as 5-(hydroxymethyl)-6- $\{[3,4,5\text{-trihydroxy-6-(hydroxymethyl)tetrahydro-2*H*-pyran-2-yl]-oxy\}$ -4,5,6,8-tetrahydro-1*H*,3*H*-pyrano[3,4-*c*]pyran-1-one (Fig. 1).

Experimental Section

General experimental procedures

The UV and IR spectra were recorded on Hitachi UV-3200 and JASCO 302-A spectrometers. ^1H , ^{13}C NMR, and 2D NMR spectra were recorded on a Bruker AM-400 spectrometer. Chemical shifts (δ) are expressed in ppm relative to TMS as the internal standard, and coupling constants (J) are given in Hz. The EI-MS and HR-EI-MS were measured on a JEOL JMS-HX-110 mass spectrometer. Silica gel 230–400 mesh (E. Merck, Darmstadt, Germany) was used for column chromatography. Diaion HP-20 ion exchange resin (Nippon Rensui Co., Mitsubishi Chemical Corporation, Tokyo, Japan) was employed for ion exchange chromatography, and silica gel plates Si 60 F₂₅₄ (E. Merck Darmstadt, Germany) for TLC. Preparative high-performance liquid chromatography (HPLC) was used for the final purification *via* recycling preparative HPLC (LC-908W-C-60, Japan Analytical Industry Co. Ltd, Tokyo, Japan) using an ODS-M-80 column ($4 \mu\text{M}$, $250 \times 200 \text{ mm}^2$; Japan Analytical Industry, Co. Ltd).

Table 1. ¹H (400 MHz) and ¹³C (100 MHz) NMR spectral data of compounds **1** and **2** (in CDCl₃; δ in ppm and *J* in Hz).

No.	1		2	
	δ _H	δ _C	δ _H	δ _C
1	–	165.4	–	165.4
2	–	–	–	–
3	4.02 (m) 4.41 (m)	65.9	4.00 (m) 4.40 (m)	65.9
4	2.48 (m)	28.8	2.47 (m)	28.6
4a	–	153.8	–	153.5
5	2.38 (m)	44.2	2.61 (m)	41.2
6	3.69 (m)	77.8	5.01 (d, 1.5)	98.7
7	–	–	–	–
8	4.25 (d, 16.0) 4.38 (m)	67.6	4.27 (d, 16.0) 4.37 (m)	66.8
8a	–	125.0	–	125.0
9	3.71 (m)	60.5	3.71 (m)	60.7
10	3.78 (dd, 7.0, 11.0) 3.64 (dd, 4.5, 11.0)	63.3	–	–
1'	–	–	5.90 (d, 7.4)	103.9
2'	–	–	3.91 (m)	73.0
3'	–	–	3.70 (m)	77.6
4'	–	–	3.53 (m)	71.6
5'	–	–	3.65 (m)	78.1
6'	–	–	3.62 (dd, 4.1, 11.0) 3.72 (dd, 5.5, 11.0)	62.8

Plant material

The whole plant material of *E. verticillatum* (Gentianaceae) was collected from Thatta region (Sindh, Pakistan) and identified by Prof. Dr. Surraiya Khatoon, Plant Taxonomist, Department of Botany, University of Karachi, Karachi, Pakistan, where a voucher specimen (No. 15013) has been deposited in the herbarium.

Extraction and isolation

The shade-dried plant material (30 kg) was extracted with MeOH (3 × 1 L) at r.t. The residue from the methanolic extract was suspended in water and successively extracted with *n*-hexane, CHCl₃, EtOAc, and *n*-BuOH. The *n*-BuOH-soluble fraction (130 g) was subjected to chromatography over a Diaion HP-20 column eluting with mixtures of MeOH and water in decreasing order of polarity. The fraction eluted

with MeOH-H₂O (3 : 1) was rechromatographed over silica gel and eluted with mixtures of CH₂Cl₂ and MeOH in increasing order of polarity. The fraction eluted with CH₂Cl₂-MeOH (9.6 : 0.4) was a binary mixture of compounds which was separated by preparative HPLC eluting with MeOH-H₂O (1 : 1). Compound **1** was obtained as a colorless gummy solid (18 mg; *t*_R = 22 min). Compound **2** was also obtained as a colorless gummy solid (15 mg; *t*_R = 41 min).

Enzymatic hydrolysis of enicostemin B (2)

Compound **2** (4 mg) was dissolved in H₂O (2 mL), to which β-glycosidase from almond (To-yobo, Japan) (2 mg) had been added, and the solution was kept at 37 °C for 22 h. After addition of H₂O (2 mL), the solution was extracted with EtOAc (5 mL). The residue recovered from the organic phase could not be worked up due to paucity of material. The aqueous layer was concentrated *in vacuo*, and the residue was purified by column chromatography over silica gel eluting with CHCl₃-MeOH with an increasing amount of MeOH to give D-glucose (2 mg) which was identified by co-TLC over silica gel with an authentic sample [solvent: *n*-BuOH-Me₂CO/H₂O (4 : 5 : 1, *t*_R = 0.35), [α]_D²⁰ = +50.1 (*c* = 0.1, H₂O)].

Enicostemin A (1)

Colorless gummy solid. – [α]_D²⁰ = –155 (*c* = 0.02, MeOH). – UV (CHCl₃): λ_{max} = 240 (4.32) nm. – IR (KBr) ν_{max}: = 3400 (OH), 1701 (ester) and 1635 cm^{–1} (conjugated C=C). – HRMS ((+)-EI): *m/z* = 214.0841 [M]⁺ (calcd. 214.0838 for C₁₀H₁₄O₅). – EIMS: *m/z* (rel. int., %) = 214 (22) [M]⁺, 196 (15), 183 (32), 178 (21), 165 (19), 153 (100). – ¹H and ¹³C NMR spectral data: see Table 1.

Enicostemin B (2)

Colorless gummy solid. – [α]_D²⁰ = –118 (*c* = 0.02, MeOH). – UV (CHCl₃): λ_{max} = 240 (4.32) nm. – IR (KBr) ν_{max}: = 3400 (OH), 1701 (ester) and 1635 cm^{–1} (conjugated C=C). – Negative HRMS ((–)-FAB): *m/z* = 361.1134 [M–H][–] (calcd. 361.1129 for C₁₅H₂₁O₁₀). – ¹H and ¹³C NMR spectral data: see Table 1.

- [1] S.I. Ali, M. Qaise, *Flora of Pakistan*, Vol. 197, Department of Botany, University of Karachi, Ferozsons Publishers, Karachi, **1995**, p. 3.
- [2] E. Nasir, S.I. Ali, *Flora of West Pakistan*, Department of Botany, University of Karachi, Ferozsons Publishers, Karachi, **1972**, p. 533.
- [3] E. Jahan, S. Perveen, A. Malik, *J. Asian Nat. Prod. Res.* **2009**, *11*, 257.
- [4] S. Ghosal, R. K. Chaudhur, M. P. Tiwari, A. K. Singh, F. W. Wehrli, *Tetrahedron Lett.* **1974**, *5*, 403.
- [5] Y.L. Li, J. Li, N.L. Wang, X. S. Yao, *Molecules* **2008**, *13*, 1931.
- [6] W.G. van der Sluis, R. P. Labadie, *Planta Med.* **1981**, *41*, 150.
- [7] D. Wang, M. Xu, H. T. Zhu, K. K. Chen, Y. Zhang, C. R. Yang, *Bioorg. Med. Chem. Lett.* **2007**, *17*, 3195.
- [8] A. I. El-Sedway, M. Hattori, K. Kobashi, T. Namba, *Chem. Pharm. Bull.* **1989**, *37*, 2435.
- [9] W. Changzeng, Y. Dequan, *Phytochemistry* **1997**, *45*, 1483.