

Die Selten-Erd-Metall(III)-Fluorid-Oxoselenate(IV) $MF[SeO_3]$ ($M = Y, Ho - Lu$) im $YF[SeO_3]$ -Typ

The Rare-Earth Metal(III) Fluoride Oxoselenates(IV) $MF[SeO_3]$ ($M = Y, Ho-Lu$)
with $YF[SeO_3]$ -type Structure

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$YF[SeO_3]$ -type rare-earth metal(III) fluoride oxoselenates(IV) $MF[SeO_3]$ ($M = Y, Ho-Lu$) crystallize monoclinically in space group $P2_1/c$ (no. 14) with $Z = 4$. Obeying the lanthanoid contraction, the lattice parameters decrease successively from $a = 657.65(7)$, $b = 689.71(7)$, $c = 717.28(7)$ pm and $\beta = 99.036(5)^\circ$ for $YF[SeO_3]$ at 298 K to $a = 648.39(6)$, $b = 681.28(7)$, $c = 705.81(7)$ pm and $\beta = 98.657(5)^\circ$ for $LuF[SeO_3]$ at 100 K (LT- $LuF[SeO_3]$, LT \equiv Low Temperature). The M^{3+} cations are occupying the general site $4e$. Contrary to the triclinic structure of RT- $LuF[SeO_3]$ (RT \equiv Room Temperature; CN(Lu^{3+}) = 7) the higher coordination number *eight* is achieved for the M^{3+} cations in all $YF[SeO_3]$ -type compounds. This results in $[MO_6F_2]^{11-}$ polyhedra ($d(M-O) = 228 - 243/225 - 239$ plus $263/258$ pm, $d(M-F) = 219/216$ pm, $M = Y/Lu$), which are connected *via* common O \cdots O edges to form infinite chains $\frac{1}{\infty}\{[MO_{4/2}^e O_{2/1}^t F_{2/1}^t]^{7-}\}$ (e \equiv edge-sharing, t \equiv terminal) running along [010]. Neighboring chains share common O3 \cdots O3 and O3 \cdots F edges generating $\frac{2}{\infty}\{[M(O3)_{3/3}(O2)_{2/2}(O1)_{1/1}F_{2/2}]^{4-}\}$ sheets parallel to the (100) plane. Finally, these $\frac{2}{\infty}\{[MO_3F]^{4-}\}$ sheets are interconnected by Se^{4+} cations, which appear in isolated ψ^1 -tetrahedral $[SeO_3]^{2-}$ anions ($d(Se-O) = 167 - 175$ pm). For the synthesis of the $YF[SeO_3]$ -type rare-earth metal(III) fluoride oxoselenates(IV) $MF[SeO_3]$ ($M = Y, Ho-Lu$), the rare-earth metal sesquioxides (M_2O_3) and trifluorides (MF_3), and selenium dioxide (SeO_2) in molar ratios of 1 : 1 : 3 with the fluxing agent CsBr were reacted within five days at 700 – 750 °C in evacuated graphitized silica ampoules.

Key words: Rare-Earth Metals, Fluorides, Oxoselenates(IV), Crystal Structures