

# Beiträge zur Chemie des Phosphors, 242 [1]

## Tri-*tert*-butylcyclotriphosphan-monoxid, (P<sup>*t*</sup>Bu)<sub>3</sub>O – Bildung und <sup>31</sup>P-NMR-spektroskopische Strukturbestimmung

Contributions to the Chemistry of Phosphorus, 242 [1]

Tri-*tert*-butylcyclotriphosphane Monoxide, (P<sup>*t*</sup>Bu)<sub>3</sub>O – Formation and Structure Determination by <sup>31</sup>P NMR Spectroscopy

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Tri-*tert*-butylcyclotriphosphane Monoxide, 1,2,3-Tri-*tert*-butyl-1-oxocyclotriphosphane, Tetra-*tert*-butylcyclotetraphosphane Monoxide, Tetra-*tert*-butylcyclotetraphosphane Dioxide, 1,2,3,4-Tetra-*tert*-butyl-1,2-dioxocyclotetraphosphane

The partially oxidized cyclotriphosphane (P<sup>*t*</sup>Bu)<sub>3</sub>O (1,2,3-tri-*tert*-butyl-1-oxocyclotriphosphane) (**1**) has been obtained by [2+1] cyclocondensation of K(Bu<sup>*t*</sup>)P–P(Bu<sup>*t*</sup>)K with Bu<sup>*t*</sup>P(O)Cl<sub>2</sub> in toluene at –78 °C. The cyclophosphanes (P<sup>*t*</sup>Bu)<sub>4</sub> and (P<sup>*t*</sup>Bu)<sub>3</sub> as well as small amounts of the open-chain triphosphane [Bu<sup>*t*</sup>P(O)Cl]<sub>2</sub>P(O)Bu<sup>*t*</sup> are also formed. In contrast to the unoxidized compound (P<sup>*t*</sup>Bu)<sub>3</sub>, compound **1** is not stable at room temperature, but decomposes above –30 °C undergoing rearrangement into the cyclotetraphosphanes (P<sup>*t*</sup>Bu)<sub>4</sub>O (**2**), (P<sup>*t*</sup>Bu)<sub>4</sub>O<sub>2</sub> (**3**), and (P<sup>*t*</sup>Bu)<sub>4</sub>. The <sup>31</sup>P NMR parameters of **1** and **3** are reported and discussed.