

Bis(*n*-pentanethiolato)mercury(II), Hg(SC₅H₁₁)₂ – Preparation, Characterization, Crystal and Molecular Structure; Reactivity towards Organic Thiols

Gerhard G. Hoffmann^{a,*}, Ingeborg Steinfatt^a, Wolfgang Brockner^b, and Volker Kaiser^c

^a Institut für Erdöl- und Erdgasforschung,
Walther-Nernst-Str. 7, D-38678 Clausthal-Zellerfeld, Germany

^b Institut für Anorganische und Analytische Chemie, Technische Universität Clausthal,
Paul-Ernst-Str. 4, D-38678 Clausthal-Zellerfeld, Germany

^c Institut für Kristallographie, RWTH Aachen, Jägerstr. 17-19, D-52056 Aachen, Germany

* Reprint requests to Dr. G. G. Hoffmann. E-mail: Gerhard.Hoffmann@ife-clausthal.de

Z. Naturforsch. **54 b**, 887–894 (1999); received March 31, 1999

Crystal structure, IR Data, Raman Data, Mass Spectrum, Bis(*n*-pentanethiolato)mercury(II)

The preparation of bis(*n*-pentanethiolato)mercury(II), Hg(SC₅H₁₁)₂, and *n*-pentanethiolato-mercury(II) iodide, IHgSC₅H₁₁, starting from HgI₂ and *n*-pentanethiol, is presented; the X-ray crystal structure determination of Hg(SC₅H₁₁)₂ reveals a layered structure of the bis(thiolate) molecule. With 178.69(1.02)^o the S-Hg-S moiety is nearly linear. The Hg-S bond length is shorter than the sum of the covalent radii of sulphur and mercury atoms indicating a primary co-ordination number of two, which is supported by the IR and Raman spectra. Assignments of $\nu_{\text{sym}}(\text{Hg-S})$ and $\nu_{\text{asym}}(\text{Hg-S})$ are given. The mass spectrum reveals the title compound to be monomeric in the gas phase. Melting and thermal decomposition have been investigated by DTA methods. The behaviour towards organic thiols is discussed. In addition, IR and Raman spectroscopic data of IHgSC₅H₁₁ are presented and discussed.