

Synthese, Kristallstruktur und Eigenschaften von Triguanidinium-tri- μ -imidocyclotriphosphat-Monohydrat und Tetraguanidinium-tetra- μ -imidocyclotetraphosphat-Tetrahydrat, [C(NH₂)₃]₃(PO₂NH)₃ · H₂O und [C(NH₂)₃]₄(PO₂NH)₄ · 4 H₂O

Synthesis, Crystal Structure, and Properties of Triguanidinium Tri- μ -imidocyclotriphosphate Monohydrate and Tetraguanidinium Tetra- μ -imidocyclotetraphosphate Tetrahydrate, [C(NH₂)₃]₃(PO₂NH)₃ · H₂O and [C(NH₂)₃]₄(PO₂NH)₄ · 4 H₂O

Norbert Stock, Barbara Jürgens^a, Wolfgang Schnick^{*,a}

Laboratorium für Anorganische Chemie der Universität, D-95440 Bayreuth

^a neue Adresse: Institut für Anorganische Chemie, Ludwig-Maximilians-Universität, Meiserstr. 1, D-80333 München

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Trimetaphosphimate, Tetrametaphosphimate, Crystal Structure, Thermal Properties, IR Data

Coarsely crystalline [C(NH₂)₃]₃(PO₂NH)₃ · H₂O (**1**) and [C(NH₂)₃]₄(PO₂NH)₄ · 4 H₂O (**2**) have been obtained by addition of [C(NH₂)₃]₂CO₃ to a freshly prepared solution of H₃(PO₂NH)₃ or to a suspension of H₄(PO₂NH)₄ · 2 H₂O, respectively, followed by diffusion controlled addition of acetone. The crystal structures of **1** and **2** have been determined by single crystal X-ray methods ([C(NH₂)₃]₃(PO₂NH)₃ · H₂O: Pbc_a; $a = 1565.6(2)$, $b = 1068.31(6)$, $c = 2091.8(2)$ pm, $Z = 8$; [C(NH₂)₃]₄(PO₂NH)₄ · 4 H₂O: Pbc_n; $a = 1739.64(10)$, $b = 1084.19(6)$, $c = 1334.96(11)$ pm, $Z = 4$). The P₃N₃ and P₄N₄ rings of the anions exhibit the chair conformation. In **1** two cyclic anions are interconnected into pairs via hydrogen bonds. In **2** columns along [001] are formed by hydrogen bonds between the anions. Hydrogen bonding through water molecules and C(NH₂)₃⁺ ions results in a three-dimensional network in **2**. Complete dehydration is achieved below 160 or 180 °C for **1** or **2**, respectively. Heating to 700 °C causes condensation reactions with evolution of NH₃ and H₂O yielding X-ray amorphous products.

* Sonderdruckanforderungen an Prof. Dr. W. Schnick. E-mail: wsc@anorg.chemie.uni-muenchen.de