

# Darstellung, $^{11}\text{B}$ - und $^{19}\text{F}$ -NMR-Spektren des Monofluoropentaiodo-*closo*-hexaboratanions sowie Kristallstruktur von $(\text{CH}_2\text{Py}_2)[\text{B}_6\text{FI}_5]$

Synthesis,  $^{11}\text{B}$  and  $^{19}\text{F}$  NMR Spectra of the Monofluoropentaiodo-*closo*-hexaborate Anion and the Crystal Structure of  $(\text{CH}_2\text{Py}_2)[\text{B}_6\text{FI}_5]$

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Monofluoropentaiodo-*closo*-hexaborate(2-), Crystal Structure,  $^{11}\text{B}$  and  $^{19}\text{F}$  NMR Spectra

By reaction of *closo*- $[\text{B}_6\text{H}_5\text{F}]^{2-}$  in alkaline solution with excess iodine the monofluoropentaiodo-*closo*-hexaborate anion  $[\text{B}_6\text{FI}_5]^{2-}$  is formed in good yield. The crystal structure of  $(\text{CH}_2\text{Py}_2)[\text{B}_6\text{FI}_5]$  has been determined by single crystal X-ray diffraction analysis (orthorhombic, space group Pnma,  $a = 13.803(2)$ ,  $b = 11.759(2)$ ,  $c = 13.936(2)$  Å,  $Z = 4$ ). The B-F-bond length is 1.41 Å, the B-I distances range from 2.13 to 2.17 Å, the B-B distances from 1.69 to 1.76 Å. According to the  $\text{C}_{4v}$  point symmetry the  $^{11}\text{B}$  NMR spectrum of the anion exhibits three singlets at +3.8, -30.1 and -33.3 ppm with the intensity ratio 1:4:1, the  $^{19}\text{F}$  NMR spectrum one quartet at -247.6 ppm with the coupling constant  $^1J(^{19}\text{F}, ^{11}\text{B}) = 54$  Hz.

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