

cis-trans-Configuration of α -Azidochalkones

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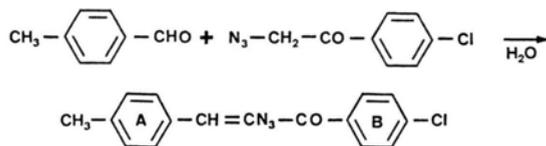
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The two stereoisomeric α -azidochalkones (1,3-diaryl-2-azidopropenones) are selectively obtainable, depending on the temperature, by condensation of aldehydes with α -azidocarbonyl compounds. For the higher melting point compound the *trans* configuration is proved by single crystal X-ray analysis of its triphenylphosphazeno derivative.

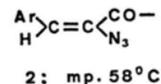
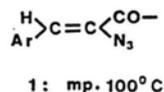
Analogous isomers may now be characterised from their ^1H NMR parameters.

Enazides reported by Smolinsky, Hassner and others [1–5] are suitable precursors to various N-heterocycles [4] and are also mechanistically of interest concerning the formation of unsaturated nitrenes [6]. Moreover, they may be of use in the electrochemical synthesis of enamine and amino-carbonyl compounds [D. Knittel, unpublished]. The preparation of unsaturated azidocarbonyls from aldehydes and α -azidocarbonyls in an one-step reaction has been discussed previously [7–9]. Condensation of *p*-tolylaldehyde with *p*-chloro- α -azidoacetophenone yields selectively either the *trans* or the *cis* form, though the existence of only one – thermodynamically stable – isomer has been postulated [2].



This selectivity can be explained by assuming a more stable intermediate for **1** but a kinetically favoured aldol for **2**. The products formed are only slightly soluble in the solvent used; thus attainment

of equilibrium is hindered and the less stable compound is produced.



An assignment of configurations from ^1H NMR spectra (see Fig. 1) is difficult because the signal of an olefinic proton is shifted by *ca.* +0.4 ppm by a *cis* CO group but by *ca.* –0.45 ppm by a *trans*

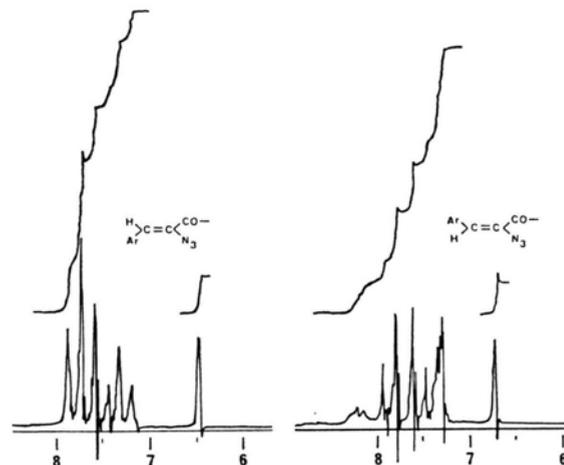
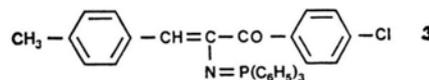


Fig. 1. ^1H NMR *vs.* TMS, CDCl_3 of **1** (left) and **2** (right).

N_3 group [3]. Since in electrochemical processes a knowledge of the geometries of substrates is important the structure of **1** was determined *via* an X-ray investigation of the triphenylphosphazeno derivative (**3**).



This is formed without changing the conformation of **1** [10] and is more stable in the X-ray beam. As can be seen from Fig. 2, **1** must be assigned the *trans* form, the torsion angle $\text{N}-\text{C}(8)-\text{C}(9)-\text{H}(9)$ being $177(1)^\circ$. However, the dihedral angle of $56(1)^\circ$ between the planes of the aromatic nuclei $\text{C}(1)$ to $\text{C}(7)$ and $\text{C}(10)$ to $\text{C}(16)$, and the torsion angle $\text{O}-\text{C}(7)-\text{C}(8)-\text{C}(9) = -151(1)^\circ$ may differ from the enazide, due to the bulky $\text{P}(\text{C}_6\text{H}_5)_3$ group.

In future it should be possible to determine by NMR spectroscopy the configuration of stereoisomers produced in analogous reactions; the olefinic signals differ by about 0.25 ppm and the signals of the aromatic protons (if present) in ring A are also different (see Fig. 1).

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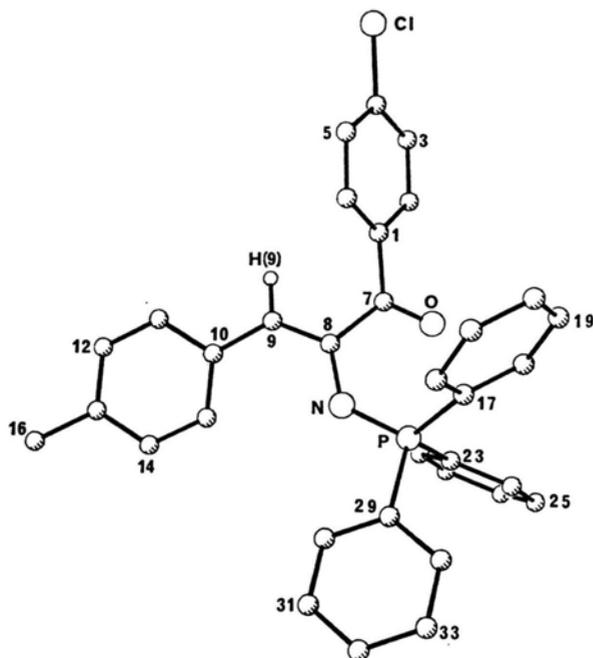


Fig. 2. A perspective view of the structure of **3** and the numbering scheme adopted.

Experimental

1-(p-Chlorophenyl)-2-azido-3-(p-tolyl)-2-propene-1-one (1): prepared by reaction of *p*-tolylaldehyde with *p*-chloro- ω -azido-acetophenone [8] at 25 °C resulting in an 80% yield of pale yellow crystals of m.p. 100 °C (dec., EtOH).

$^1\text{H NMR}$ (16% in CDCl_3/TMS): δ 2.39 (s; 3 H), 6.46 (s; 1 H), 7.50 (qc; 4 H), 7.70 (qc; 4 H).

cis-Azidochalkone (2): the same procedure was followed, except that the reaction temperature was kept below 10 °C. 70% of pure **2** (checked by NMR) were obtained, m.p. 58 °C (petrol ether).

$^1\text{H NMR}$ (as above): δ 2.23 (s; 3 H), 6.70 (s; 1 H), 7.70 (qc; 4 H), 7.2–7.4 (m; 3 H), 8.22 (m; 1 H).

1-(p-Chlorophenyl)-2-(triphenylphosphazeno)-3-(p-tolyl)-2-propene-1-one (3): prepared by refluxing **1** with slight excess of triphenylphosphine. 90% yield of orange crystals, m.p. 149 °C (benzene) were obtained.

$^1\text{H NMR}$ (CDCl_3/TMS): δ 2.30 (s; 3 H), 6.05 (s; 1 H), 6.8–8.2 (m; 23 H). A sample for X-ray analysis was obtained by recrystallisation from benzene.

X-ray Analysis of 3

Crystal data: $\text{C}_{34}\text{H}_{27}\text{ClNOP}$, $M = 532.0$, space-group $P2_1/n$, $a = 1206.9(6)$, $b = 1494.7(8)$, $c = 1604.7(11)$ pm, $\beta = 98.89(2)^\circ$, $Z = 4$, $d_{\text{calc}} = 1.235$, $d_{\text{exp}} = 1.23$ Mg m^{-3} . 2973 unique observed ($F > 3\sigma(F)$) diffractometer data, collected between $2\theta = 6$ – 46° with monochromated $\text{MoK}\alpha$ -radiation and corrected for L_p but not for extinction and absorption effects ($\mu = 0.211$ mm^{-1}), were used for solving the structure by direct methods and for anisotropic refinement by blocked cascade least-squares methods ($R = 0.052$). For H atoms (in calculated positions) a 'riding model' was employed [11]. Atomic parameters, bond lengths and angles, torsion angles and lists of observed and calculated structure amplitudes may be obtained from G. W. on request.

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