

## Number of Benzenoid Hydrocarbons

Ivan Gutman

Faculty of Science, University of Kragujevac, Yugoslavia

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The number of benzenoid hydrocarbons with  $h$  hexagons can be estimated by means of the formula  $B_h = 0.045 h^{-3/2} (5.4)^h$ . The analogous estimate for the number of catacondensed benzenoids is  $C_h = 0.049 h^{-5/4} (4.27)^h$ .

### Introduction

The problem of the enumeration of alkanes and related chemical compounds has been solved by Pólya long time ago [1, 2]. Another problem of this kind, namely the enumeration of benzenoid hydrocarbons, seems to be much more difficult and was not satisfactorily solved so far. In spite of serious attempts of a number of mathematicians [3–6], at the present moment we do not know anything better than to construct the benzenoids and then to count them. Several approaches along these lines have been elaborated [7–15], usually based on an extensive use of computers.

In the following we shall be interested in geometrically planar, simply connected benzenoids [16]. The number of such systems, possessing  $h$  hexagons will be denoted by  $B_h$ . In addition to this,  $C_h$  is the number of geometrically planar catacondensed benzenoids [16] with  $h$  hexagons. The numbers  $B_h$  and  $C_h$  are nowadays known for  $h \leq 11$  and are given in Tables 1 and 2.

For small values of  $h$ ,  $B_h$  and  $C_h$  can be obtained without difficulty. For  $h$  up to 10,  $B_h$  and  $C_h$  were first reported by Knop et al. [11, 12], whereas  $B_{11}$  and  $C_{11}$  were recently calculated by Doroslovački and Tošić [15]. The amount of computing, required for the evaluation of  $B_h$  for  $h \geq 10$  is enormous and increases very rapidly with the increasing number of hexagons. Therefore, even when quite powerful computing machines have been employed, the enumeration procedure could not exceed  $h = 10$  [11, 14] and  $h = 11$  [15].

These difficulties motivated us to develop approximate expressions for  $B_h$  and  $C_h$  which enable the estimation of these numbers for large (greater than eleven) values of  $h$ .

Asymptotic expressions for the estimation of the number of combinatorial objects of certain types are often met in the theory of enumeration [17]. Already Pólya [1] deduced such a formula for the number of alkanes. In a great number of cases [17], the asymptotes have the form

$$X_n \sim a n^p b^n; \quad n \rightarrow \infty, \quad (1)$$

where  $a$  and  $b$  are some constants and the exponent  $p$  is a rational number.

Reprint requests to Prof. Dr. Ivan Gutman, Faculty of Science, P.O. Box 60, YU-34000 Kragujevac, Yugoslavia.

In particular, Harary and Read [5] showed that for large values of  $h$ ,

$$H_h \sim \sqrt{\frac{5}{4}} \frac{(2h-1)!}{(h-1)!(h+2)!} \left(\frac{5}{4}\right)^h, \quad (2)$$

where  $H_h$  counts the catacondensed benzenoids (both geometrically planar and non-planar), with  $h$  hexagons. Using the Stirling approximation, we can transform (2) into

$$H_h \sim \sqrt{\frac{5}{16\pi}} h^{-5/2} 5^h, \quad (3)$$

which is just a special case of (1).

### An Approximate Asymptote for $C_h$

Bearing in mind the result (3), we considered the formula

$$C_h \sim a h^p b^h, \quad (4)$$

which is expected to hold for sufficiently large values of  $h$ . In order to determine the exponent  $p$  we have calculated the expression

$$b_h = \left(\frac{h}{h+1}\right)^p \frac{C_{h+1}}{C_h}$$

for the known  $C_h$ 's (see Table 1). If  $C_h$  behaves according to (4), then for a properly chosen  $p$  the sequence  $b_1, b_2, b_3, \dots$  will rapidly converge to its limit value  $b$ . By varying  $p$  we found that the best convergence occurs for  $p = -5/4$ . (As a matter of fact, the choice  $p = -1.25$  is better than  $p = -1.24$  or  $p = 1.26$ .) The last calculated members of the sequence  $b_h$ , for  $p = -5/4$  are given as

$h$	$b_h$
7	4.116
8	4.198
9	4.26887
10	4.26895

from which we conclude that the limiting value of  $b_h$  is about 4.27.

It remains to determine the parameter  $a$  as the limit of the sequence  $a_1, a_2, a_3, \dots$ , where

$$a_h = C_h / (h^p b^h).$$

The fact that for  $p = -5/4$  and  $b = 4.27$ ,

$h$	$a_h$
8	0.0500
9	0.049187
10	0.049174
11	0.049162

implies  $a = 0.049$ .

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Table 1. Exact and estimated values for the number  $C_h$  of geometrically planar catacondensed benzenoid systems with  $h$  hexagons.

Table 2. Exact and estimated values for the number  $B_h$  of geometrically planar, simply connected benzenoid systems with  $h$  hexagons.

Table 1			Table 2		
$h$	$C_h$	Estimate (5)	$h$	$B_h$	Estimate (6)
1	1	0	1	1	0
2	1	0	2	1	0
3	2	1	3	3	1
4	5	3	4	7	5
5	12	9	5	22	18
6	36	32	6	81	76
7	118	111	7	331	325
8	411	402	8	1 435	1 438
9	1 489	1 483	9	6 505	6 507
10	5 572	5 552	10	30 086	30 002
11	21 115	21 046	11	141 229	140 428
12		80 604	12		665 527
13		311 408	13		3 187 251
14		1 212 066	14		15 400 439
15		4 747 884	15		74 986 317

Thus we arrived at the approximate asymptotic expression

$$C_h \sim 0.049 h^{-5/4} (4.27)^h, \quad (5)$$

whose quality can be seen from the data given in Table 1. In Table 1 we also presented the predicted (approximate) values of  $C_{12}$ ,  $C_{13}$ ,  $C_{14}$  and  $C_{15}$ .

#### An Approximate Asymptote for $B_h$

In the case of  $B_h$ , a completely analogous variational procedure gave the optimal value  $-1.47$  for the exponent  $p$ , which is satisfactorily close to the adopted value  $-3/2$ . The choice  $p = -3/2$  leads then to the numbers

$h$	$b_h$	$a_h$	
7	5.297		which imply
8	5.409	0.0449	
9	5.41693	0.044985	$B_h \sim 0.045 h^{-3/2} (5.4)^h$ .
10	5.41562	0.045126	(6)
11	—	0.045224	

The exact  $B_h$  values as well as those calculated by means of (6) are collected in Table 2, together with the estimates for  $B_{12}$ ,  $B_{13}$ ,  $B_{14}$  and  $B_{15}$ . Formula (6) seems to be somewhat less accurate than (5).

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 [16] A benzenoid system is geometrically planar if its non-adjacent (regular) hexagons do not overlap. A benzenoid system is simply connected if it separates the plane into an infinite region and  $h$  finite regions, all of which are (regular) hexagons. A benzenoid system is catacondensed if no three of its hexagons are mutually adjacent. More details on benzenoid systems can be found in: I. Gutman and O. E. Polansky, Mathematical Concepts in Organic Chemistry, Springer-Verlag, Berlin 1986, pp. 59–61.  
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