

Dipole Moment of Thionitrosylchloride

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The microwave spectrum of Thionitrosylchloride, NSCl, has been investigated by BEPPU, HIROTA, and MORINO¹ who determined the rotational constants, the structure and the force constants.

In the present work the dipole moment is determined. A sample of N₃S₃Cl₃ was kindly given to me by Dr. J. WEGENER of the University of Göttingen: NSCl is then obtained by heating the N₃S₃Cl₃ to 50 °C under high vacuum. The measurement has been made with a conventional 100 kHz Stark modulated spectrometer with automatic frequency markers²⁻⁴; frequency measurements are believed to be accurate within ±0.03 MHz.

The electric field was calibrated by means of the $J=1 \rightarrow J'=2$ transition of OCS using the value of 0.7152⁵ D for the dipole moment.

The measurements have been performed on the $J=0_{00} \rightarrow J'=1_{11}$ line at 45,460.22 MHz in the high field approximation and further on the $F=7/2 \rightarrow F'=7/2$ hyperfine line at 38,360.62 MHz of the $J=2_{02} \rightarrow J'=2_{11}$ transition and the $F=9/2 \rightarrow F'=9/2$ hyperfine line at

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¹ T. BEPPU, E. HIROTA, and Y. MORINO, J. Mol. Spectr. 36, 386 [1970].

Table 1. Measured and calculated slopes $\Delta\nu/E^2$ in MHz/(V/cm)².

$Jk_{-1}, k_1 \rightarrow J'k'_{-1}, k'_1$	$F \rightarrow F'$	$\Delta\nu/E^2$	
		obs.	calc.
20,2 → 21,1	7/2 → 7/2	2.212 ± 0.05	2.218
30,3 → 31,2	9/2 → 9/2	1.688 ± 0.020	1.684
	$M_J, M_I \rightarrow M'_J, M'_I$		
00,0 → 11,1 ^a	$\left. \begin{array}{l} 0,3/2 \rightarrow 0,3/2 \\ 0,1/2 \rightarrow 0,1/2 \end{array} \right\}$	1.527 ± 0.015	1.532

^a Measured in high field condition.

$ \mu_a $	= 0.57 ± 0.03
$ \mu_b $	= 1.77 ± 0.02
$ \mu_{tot} $	= 1.87 ± 0.02

Table 2. Values for the two components and for the total electric dipole moment in Debye.

38,933.92 MHz of the $J=3_{03} \rightarrow J'=3_{12}$ transition whose Stark Effect is independent of the quadrupole energy. In Table 1 the measured and calculated slopes for these lines are given, and in Table 2 the dipole moment components along the two principal axes of inertia and the total dipole moment are listed.

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² H. D. RUDOLPH, Z. Angew. Phys. 13, 401 [1961].

³ H. D. RUDOLPH and H. SEILER, Z. Naturforsch. 20 a, 1682 [1965].

⁴ U. ANDRESEN and H. DREIZLER, Z. Angew. Phys. 30, 207 [1970].

⁵ J. S. MUENTER, J. Chem. Phys. 48, 4544 [1968].

BERICHTIGUNG

Erratum. B. D. HANSON, M. MAHNIG, and LOUIS E. TOTH, Low Temperature Heat Capacities of Transition Metal Borides, Z. Naturforsch. 26 a, 739 [1971].

P. 739, column 2, line 13 ff from bottom should read: There is, however, little general agreement on the number of electrons transferred per boron atom . . . ;
p. 743, Table 3, line 3: (Mn₉Fe₁)₂B, line 6: (Mn₁Fe₉)₂B, line 13: (Co₉Ni₁)₂B;
p. 743, Table 6, line 2: Co₉₈B₀₂.

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